

## Chemistry Paper 2 SI Tz1 2012 Markscheme

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### Chemistry Paper 2 SI Tz1

Chemistry SL Paper 2 TZ1 Markscheme 2 Topics: Covalent bond , Hydrogen bond , Yield Pages: 13 (2237 words) Published: March 3, 2015

### Chemistry SL Paper 2 TZ1 Markscheme 2 - 2237 Words

8 20 2 2 2 2 o correct reactants and products; [2] correct coefficients; M2 can only be scored if M1 correct. (b) Local pollutant: carbon monoxide/CO / volatile organics/VOCs / nitrogen oxide/NO / (unburnt) hydrocarbons; Do not accept methane/CH<sub>4</sub>, ethane/C<sub>2</sub>H<sub>6</sub>, propane/C<sub>3</sub>H<sub>8</sub> or butane/C<sub>4</sub>H<sub>10</sub>. Global pollutant: nitrogen oxide/NO / carbon ...

### MARKSCHEME

- 2 - Section A Answer all questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid, HOOC-COOH·xH<sub>2</sub>O, by titrating a known mass of the acid with a 0.100 mol dm<sup>-3</sup> solution of NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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[2] (b) Marit arranged the values she found in Table 12 into an energy cycle. C<sub>2</sub>H<sub>4</sub>(g) + H<sub>2</sub>(g) C<sub>2</sub>H<sub>6</sub>(g) 3 O 2 2CO<sub>2</sub>(g) + 3H<sub>2</sub>O - 1 4 1 1 k J m o l - 1 - 2 8 6 k J m o l - 1 - 1 5 6 0 k J m o l - 1 (hydrogenation) (l) Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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(CH<sub>3</sub>)<sub>2</sub>CHCH<sub>2</sub>OH; (CH<sub>3</sub>)<sub>3</sub>COH; [2 max] Accept condensed or full structural formulas. Penalise missing H atoms or incorrect bonds (such as C-HO, C-H 2C) only once in the whole paper. (c) C H CHOCH [O] C H COCH H O<sub>25</sub> 3 25 3 2+→ +; [1] Accept condensed or full structural formulas. Accept [O] on top of the arrow. Do not accept equation without H 2O.

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