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Modern Chemistry Chapter 12 Solutions Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a

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CHAPTER 12 REVIEW Solutions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Solid CaCl 2 does not conduct electricity. Explain why it is considered to be an electrolyte. CaCl 2 is an ionic solid. In the crystal form, its ions are locked in position. Dissolving

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Modern Chemistry Chapter 12. STUDY. PLAY. soluble. Capable of being dissolved. solution. A homogeneous mixture of two or more substances in a single phase. solvent. In a solution, the substance in which the solute dissolves. suspension. Mixture in which a substance will not dissolve in another and quickly separates out if left to stand.

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Chemistry Principles of Modern Chemistry The thermite process (see Fig. 12.16) is used for welding railway track together. In this reaction, aluminum reducesiron(III) oxide to metallic iron: $2 \text{Al} (s) + \text{Fe}_2 \text{O}_3 (s) \rightarrow 2 \text{Fe} (s) + \text{Al}_2 \text{O}_3 (s)$ Igniting a small charge of barium peroxide mixed with aluminum triggers the reaction of a mixture of a luminium powder and iron(III) oxide; the ...

The thermite process (see Fig. 12.16) is used for welding ...

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