

## Thermochemistry Guided Practice Problems

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### Thermochemistry Guided Practice Problems

Thermochemistry Practice Problems (Ch. 6) 1. Consider 2 metals, A and B, each having a mass of 100 g and an initial temperature of 20 °C. The specific heat of A is larger than that of B. Under the same heating conditions, which metal would take longer to reach 21 °C? Explain your reasoning. 2.

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### Thermochemistry Guided Practice Problems

Thermochemistry practice problems 1) How can energy be transferred to or from a system? A) Energy can only be transferred as potential energy being converted to kinetic energy. B) Energy can be transferred only as heat. Ene can be transferred onl as work. D) Energy can be transferred as heat and/or work.

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### Thermochemistry Guided Practice Problems

Thermochemistry Exam1 and Problem Solutions 1. Which ones of the following reactions are endothermic in other words  $\Delta H$  is positive? I.  $\text{H}_2\text{O}(\text{l}) + 10,5\text{kcal} \rightarrow \text{H}_2\text{O}(\text{g})$   $\Delta H_1$  II.  $2\text{NH}_3 + 22\text{kcal}$

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### Chapter 17 Thermochemistry Practice Problems [PDF]

Calorimetry Practice Problems 1. How much energy is needed to change the temperature of 50.0 g of water by 15.0°C? 2. How many grams of water can be heated from 20.0 °C to 75°C using 12500.0 Joules? 3. What is the final temperature after 840 Joules is absorbed by 10.0g of water at 25.0°C? 4. The heat capacity of aluminum is 0.900 J/g°C. a.

### Calorimetry Practice Problems

Do practice problems: When carbon disulfide is formed from its elements, heat is absorbed. Calculate the amount of heat (in kJ) absorbed when 5.66g of carbon disulfide is formed.  $\text{C}(\text{s}) + 2\text{S}(\text{s}) \rightarrow \text{CS}_2(\text{l})$   $\Delta H = 89.3 \text{ kJ}$ . The production of iron and carbon dioxide from iron (III) oxide and carbon monoxide is an exothermic reaction.

### Chapter 11: Thermochemistry-Heat and Chemical Change

Thermochemistry Practice Problems. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Altrum. Terms in this set (22)-40. To start a heat pack, 20kJ of work had to be done on it first. Once started, the chemical reaction in the heat pack released 60 kJ of heat. Calculate the total energy change.

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